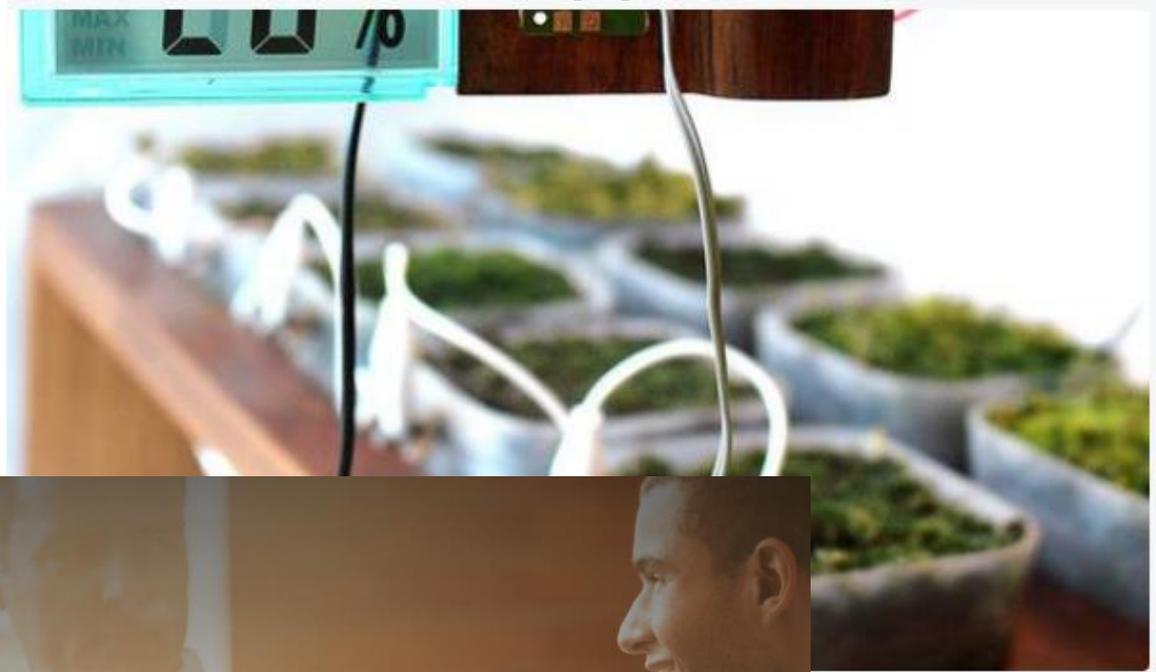


Photosynthetic bio electrochemical systems (photosynthetic-BES)

Possible areas of application



September 2018

Paolo pb346@cam.ac.uk

@ Department of Biochemistry (University of Cambridge, UK)

Photosynthetic organisms

(few example)



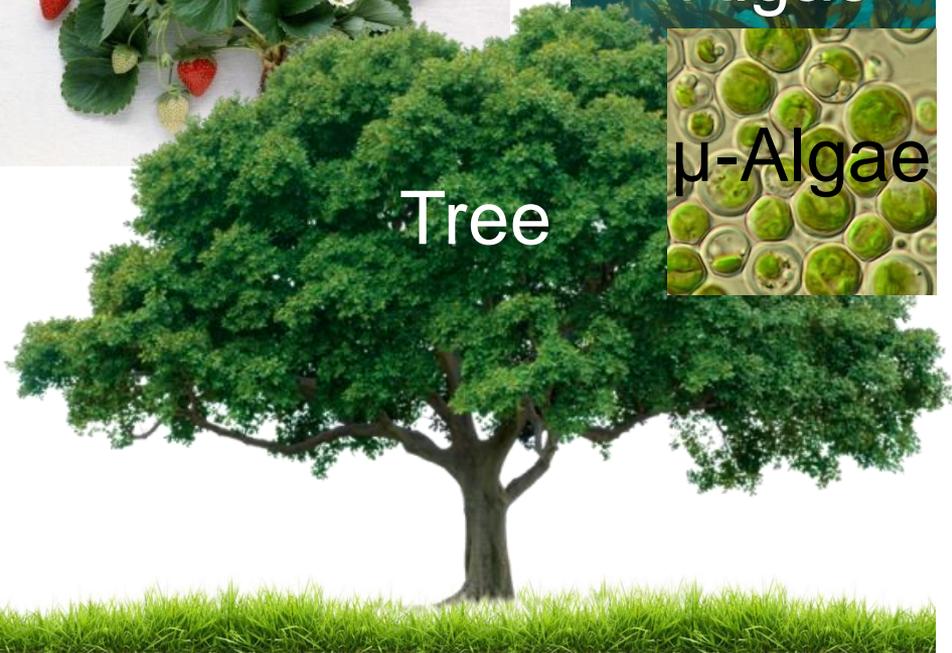
Plant



Algae



μ-Algae



Tree

Photosynthetic organisms

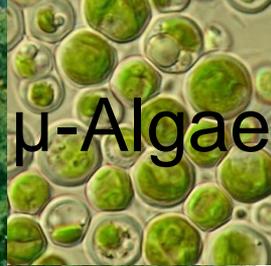
(few example)



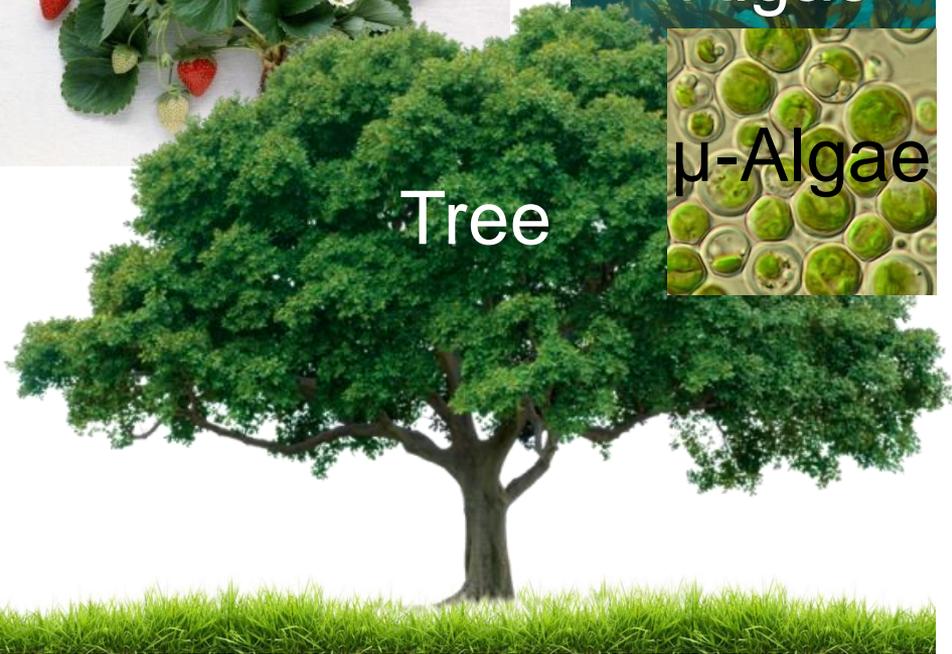
Plant



Algae



μ -Algae



Tree

Products



Fibers/biomass



Fuel(s)



Food(s)



Photosynthetic organisms

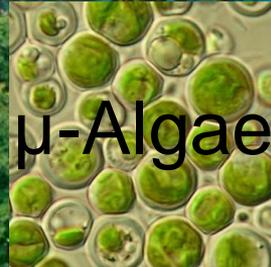
(few example)



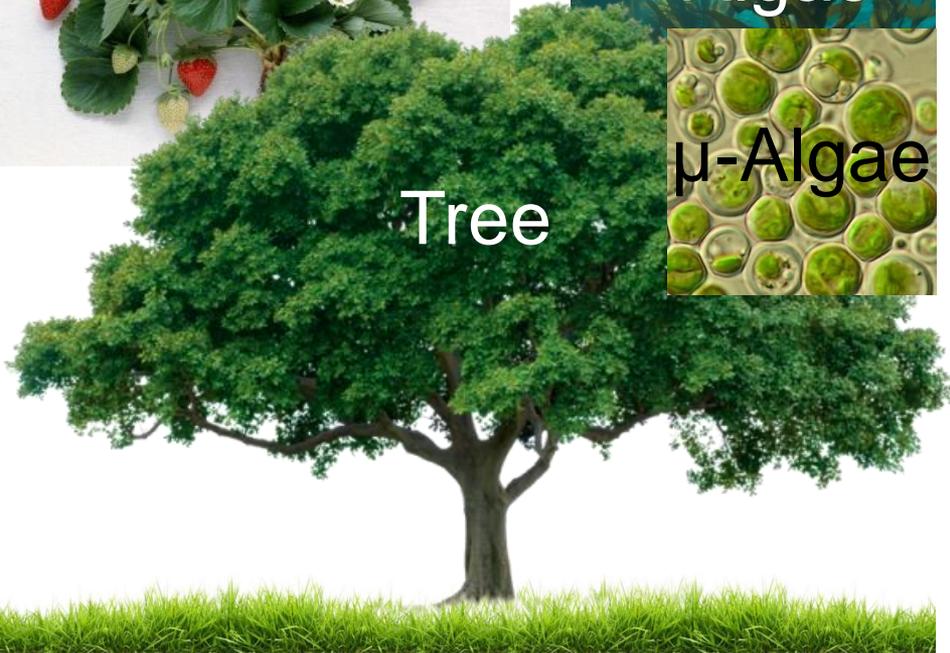
Plant



Algae



μ -Algae



Tree

Products



Fibers/biomass



Fuel(s)



Food(s)



Electricity



Photosynthetic organisms

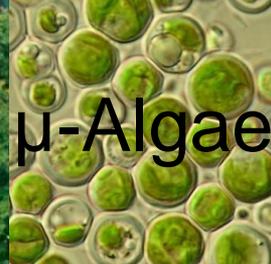
(few example)



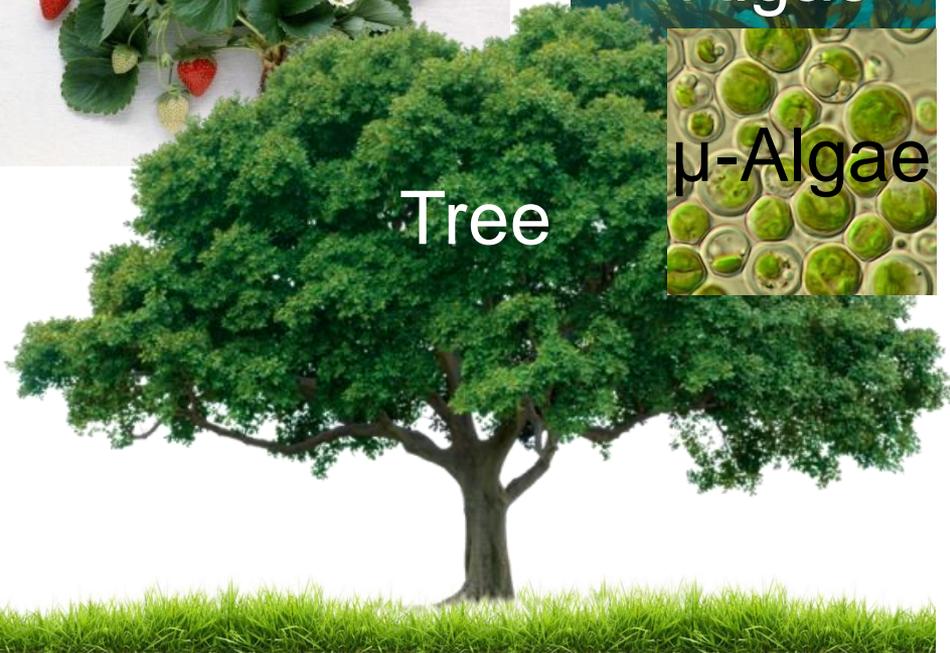
Plant



Algae



μ -Algae



Tree

Products



Fibers/biomass



Fuel(s)



Food(s)



Electricity



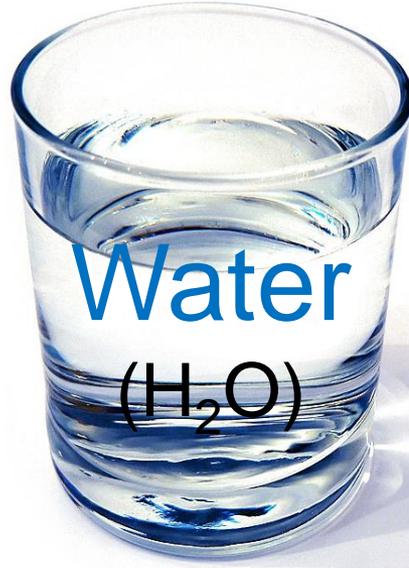
How does it work?

Photosynthetic organisms generate electricity: how does it work?

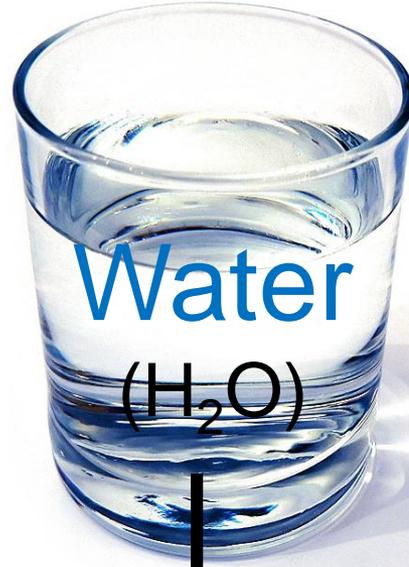
Photosynthetic organisms generate electricity: how does it work?



Photosynthetic organisms generate electricity: how does it work?



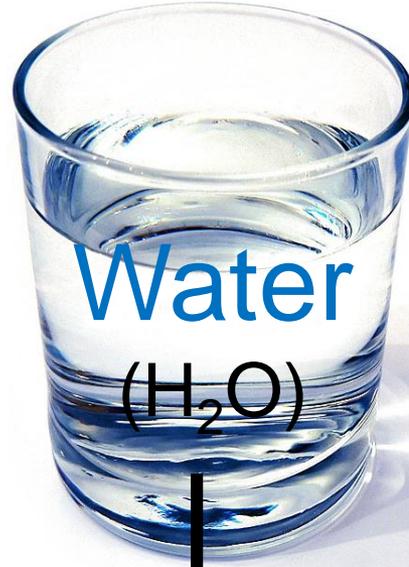
Photosynthetic organisms generate electricity: how does it work?



2H^+ (2 protons)

+ $\frac{1}{2} \text{O}_2$ ($\frac{1}{2}$ molecular oxygen)

Photosynthetic organisms generate electricity: how does it work?



2H⁺ (2 protons)

+ 1/2 O₂ (1/2 molecular oxygen)

+2e⁻ (2 electrons)

How many **electrons** could be generated from a glass of water (ca. 180mL)



$2200 \text{ mAh} = 7920 \text{ C}$



How many **electrons** could be generated from a glass of water (ca. 180mL)



$2200 \text{ mAh} = 7920 \text{ C}$

Glass of water = 180 mL

Water molecular weight: 18 mol/g

Water density $\sim 1 \text{ g/mL}$

Glass of water = $(180 / 18) = 10 \text{ mol of H}_2\text{O}$

10 mol of $\text{H}_2\text{O} = 20 \text{ mol of electron}^-$

Faraday(F) $\sim 96500 \text{ Coulomb}$

Glass of water = $20 \text{ mol} \times F = 1.93 \times 10^6 \text{ C}$

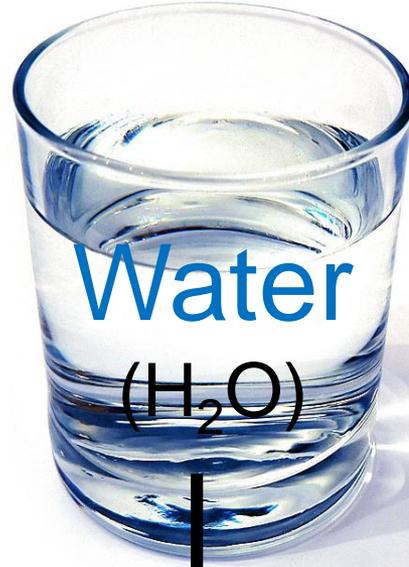
$= 1.93 \times 10^6 \text{ C} / 7920 \text{ C} = 243.7$



~ 244

Glass of water =

Photosynthetic organisms generate electricity: how does it work?

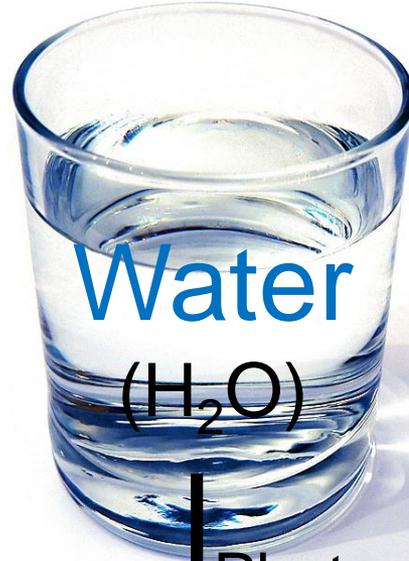


2H⁺ (2 protons)

+ 1/2 O₂ (1/2 molecular oxygen)

+2e⁻ (2 electrons)

Photosynthetic organisms generate electricity: how does it work?



+ **Light**

Photosynthetic (oxygenic) organisms

2H⁺ (2 protons)

+ **1/2 O₂** (1/2 molecular oxygen)

+ **2e⁻** (2 electrons)

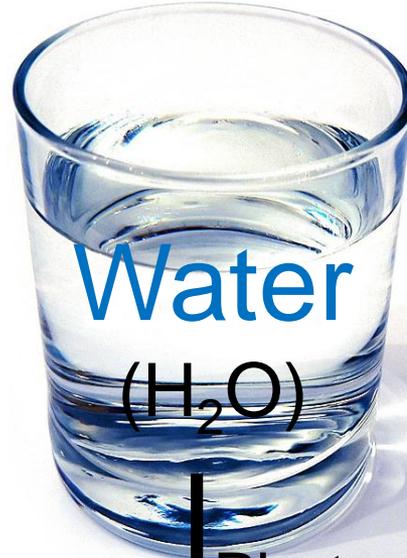


Water
photolysis

Photosynthetic organisms generate electricity: how does it work?



Water
photolysis



2H^+ (2 protons)

+ $\frac{1}{2} \text{O}_2$ ($\frac{1}{2}$ molecular oxygen)

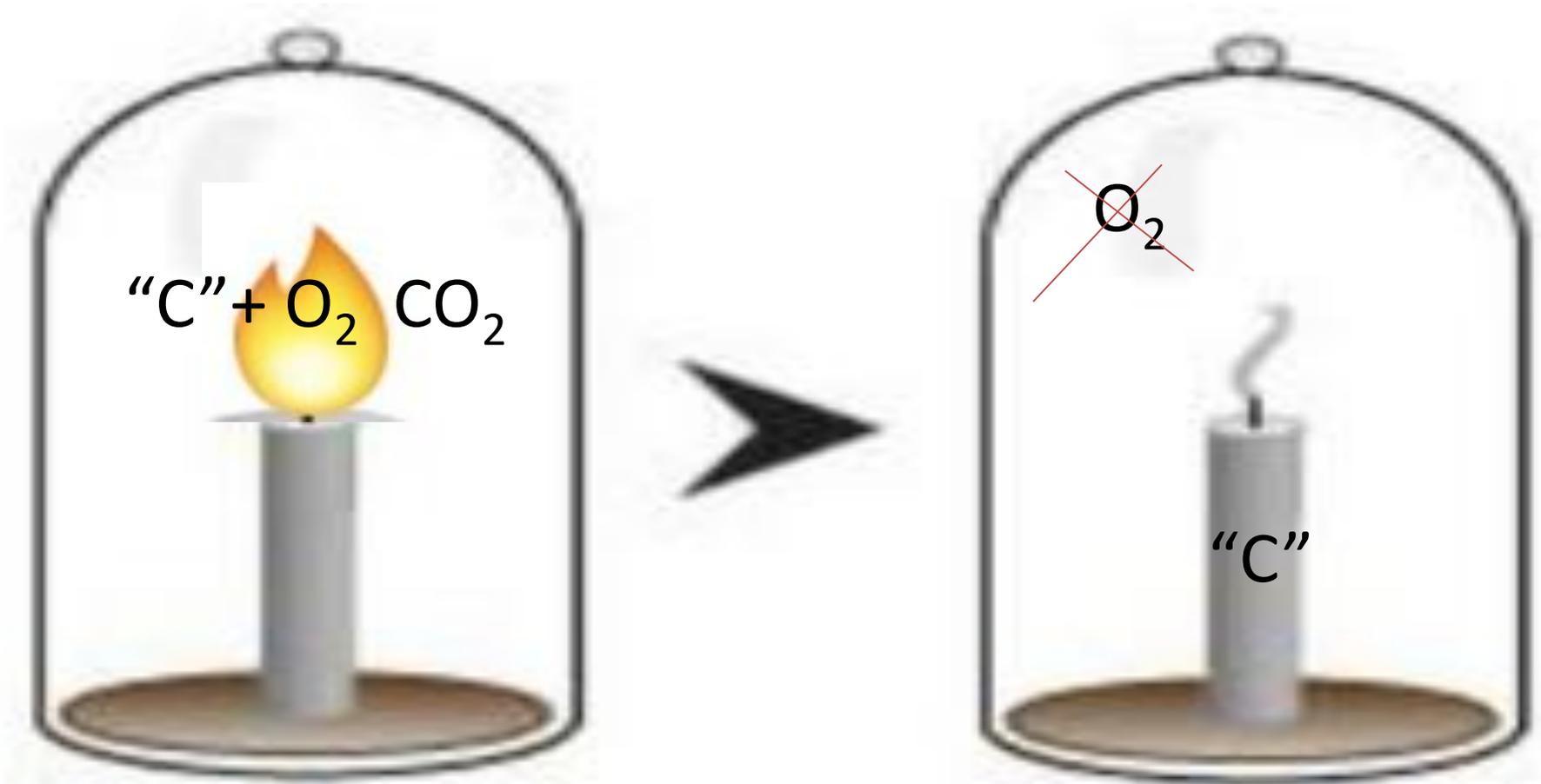
+ 2e^- (2 electrons)

+ Light

Photosynthetic (oxygenic) organisms

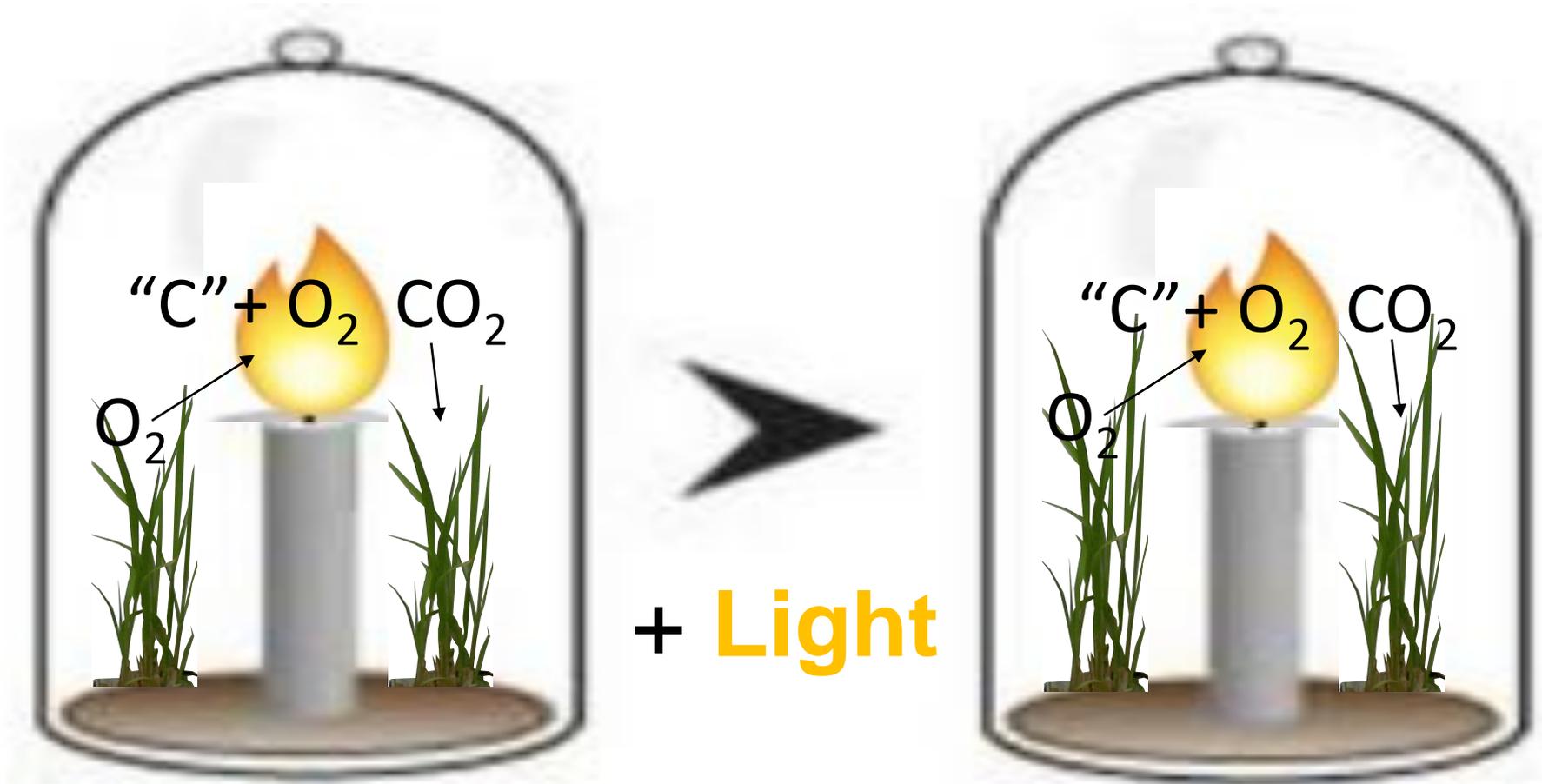
Can I prove that?

To prove that photosynthetic organisms generate **oxygen**
(and therefore **electrons**)



Priestley's experiment

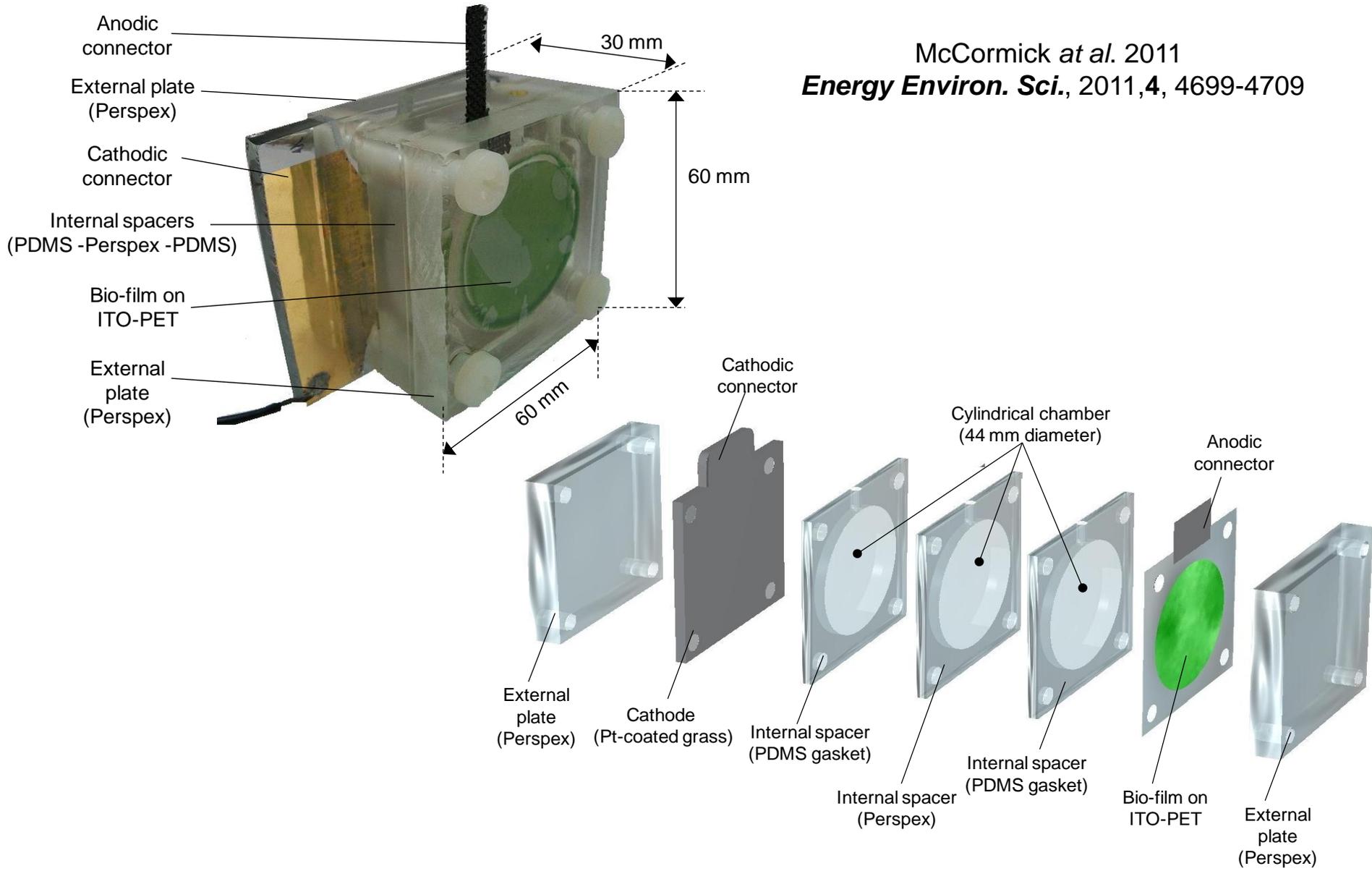
To prove that photosynthetic organisms generate **oxygen**
(and therefore **electrons**)



Priestley's experiment

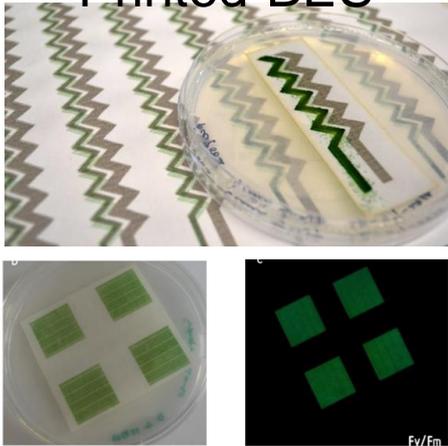
Photosynthetic bio electrochemical systems (photosynthetic-BES)

McCormick *et al.* 2011
Energy Environ. Sci., 2011, 4, 4699-4709



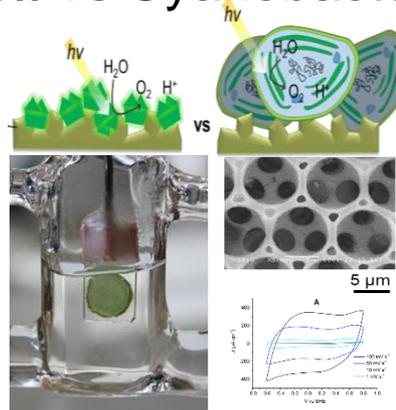
Photosynthetic bio electrochemical systems (photosynthetic-BES)

Printed-BES



DOI:10.1038/s41467-017-01084-4
Nature Communications, 2017

PSII vs Cyanobacteria



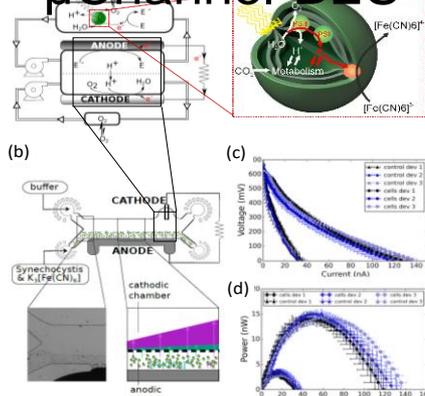
DOI: 10.1021/jacs.7b08563
J. Am. Chem. Soc., 2017

Electricity from rice



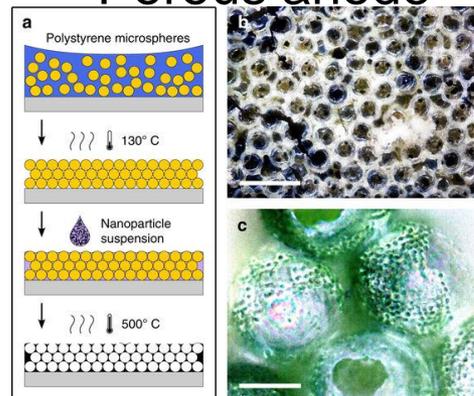
DOI: 10.1007/s00253-012-4473-6
Ap. Microb. and Biotech., 2013

μChannel-BES



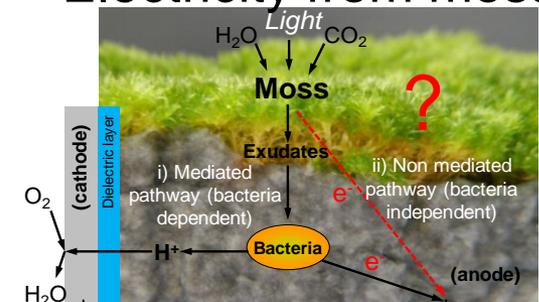
DOI:10.1038/s41560-017-0073-
Nature Energy, 2018

Porous anode



DOI: 10.1038/s41467-018-03320-x
Nature Communications, 2018

Electricity from moss



DOI: 10.1098/rsos.160249
Royal Society j. Open Science, 2016

Photosynthetic-BESs:

are they feasible for actual applications?

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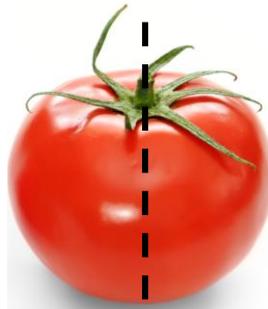
It depends from the
given conditions

Photosynthetic-BESs:

are they feasible for actual applications?

It depends from the
given conditions

Palermo (Sicily)
Summer



Glasgow (Scotland)
Winter

Photosynthetic-BES as educational toolkit

Bio-Bottlevoltaic
bio-hack electricity



iaac

Lara Allen

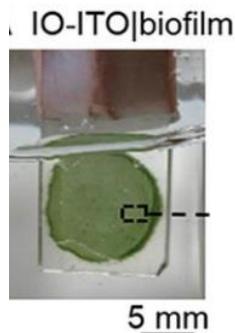


<http://www.iaacblog.com/programs/bio-bottle-voltaic-bbv/>

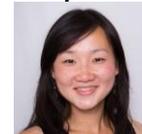
Photosynthetic-BES as tool of investigation for lab work

e.g., to investigate the mechanism of electron transport

J. Am. Chem. Soc., 2017
DOI: 10.1021/jacs.7b08563



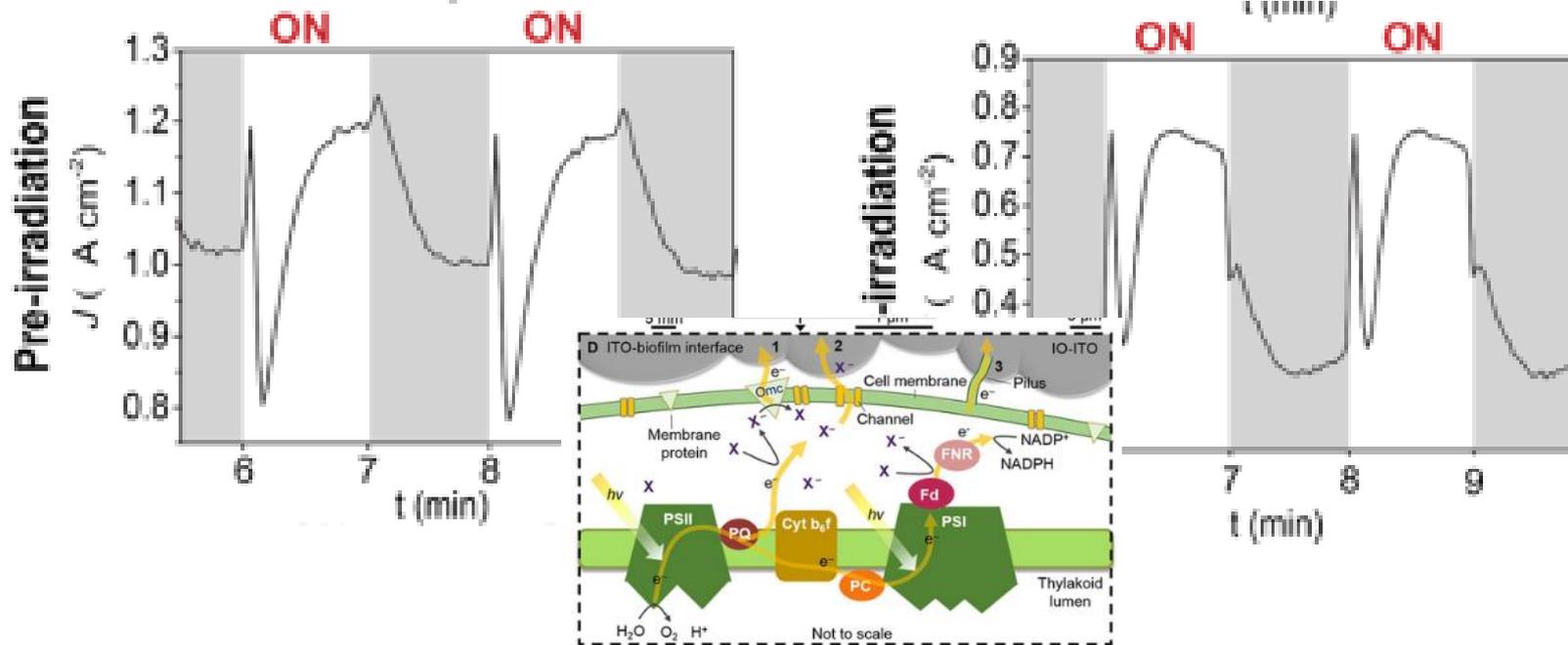
Jenny Zhang



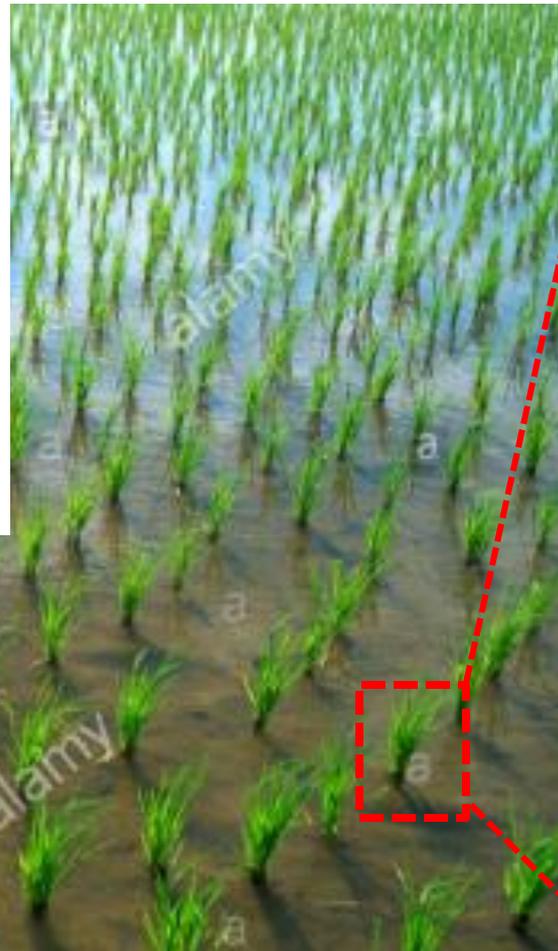
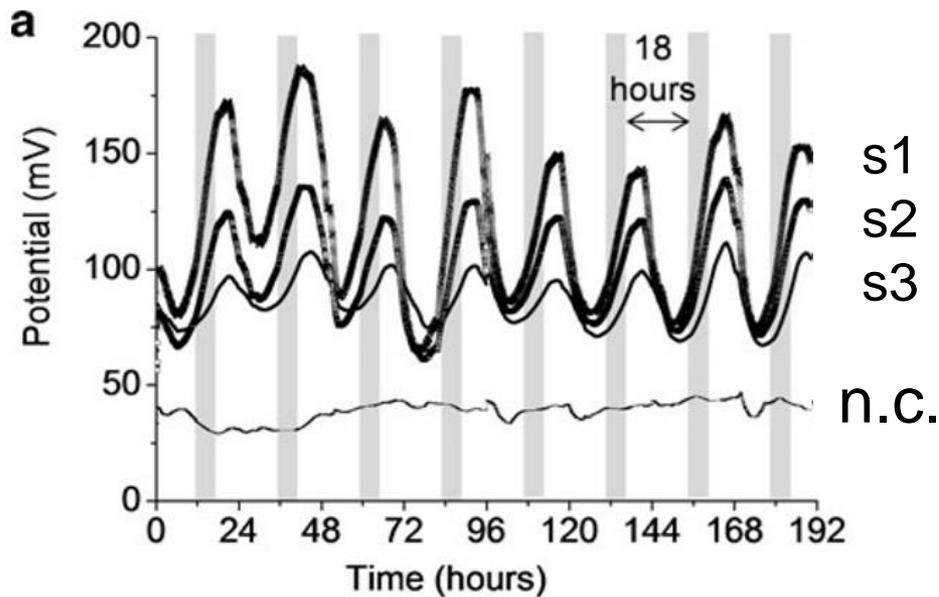
Laura Wey



Chronoamperometry



Photosynthetic-BES as tool of investigation for agricultural work



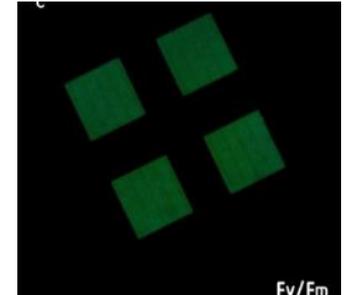
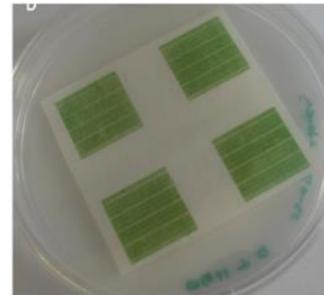
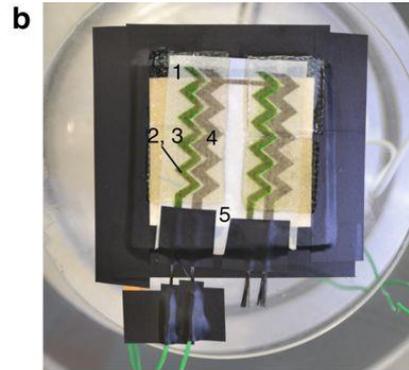
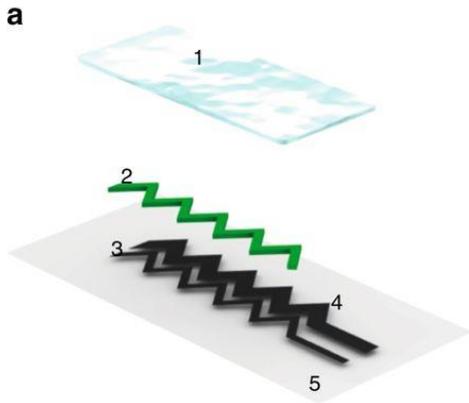
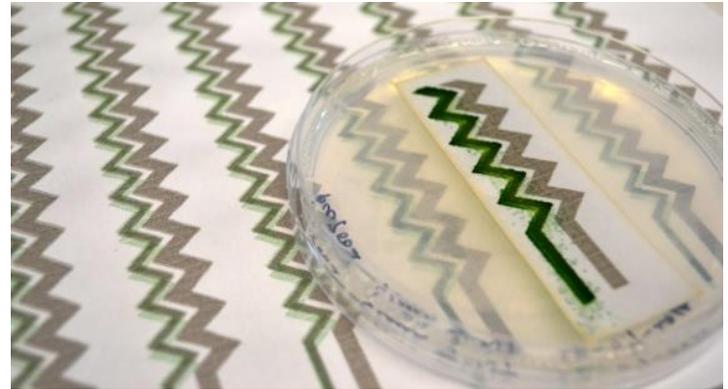
I.o.P.

Photosynthetic-BES used to run μ -chip

arm

Anand
Savanth

Emre
Ozer



Andrea
Fantuzzi

Peter
Nixon

Marin
Sawa

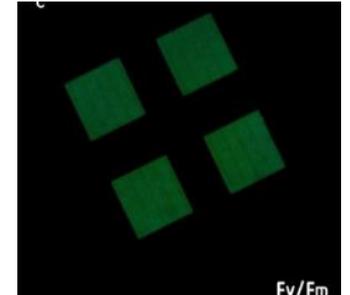
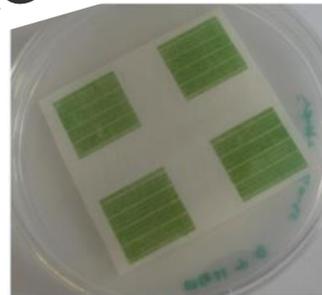
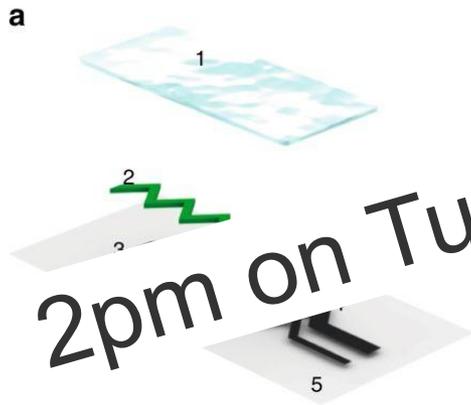


DOI:10.1038/s41467-017-01084-4
Nature Communications, 2017

Imperial College
London

Photosynthetic-BES used to run μ -chip

arm



2pm on Tue in the printed electric session

Andrea Fantuzzi

Peter Nixon

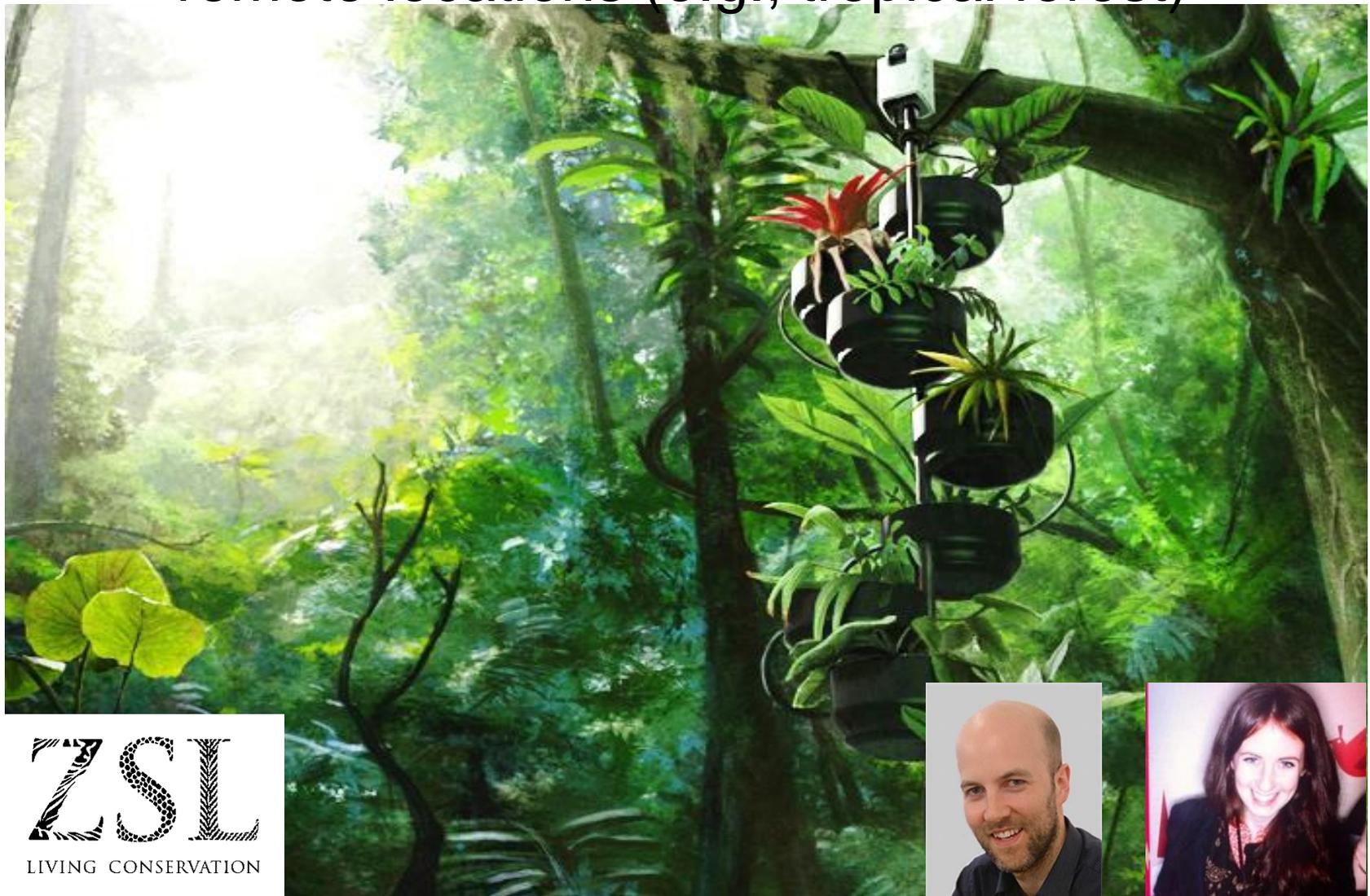
Marin Sawa



DOI:10.1038/s41467-017-01084-4
Nature Communications, 2017

Imperial College
London

Photosynthetic-BES used to run environmental sensor in remote locations (e.g., tropical forest)



Alasdair Davies Rachael Kemp

Photosynthetic-BES used to run environmental sensor in remote locations (e.g., tropical forest)

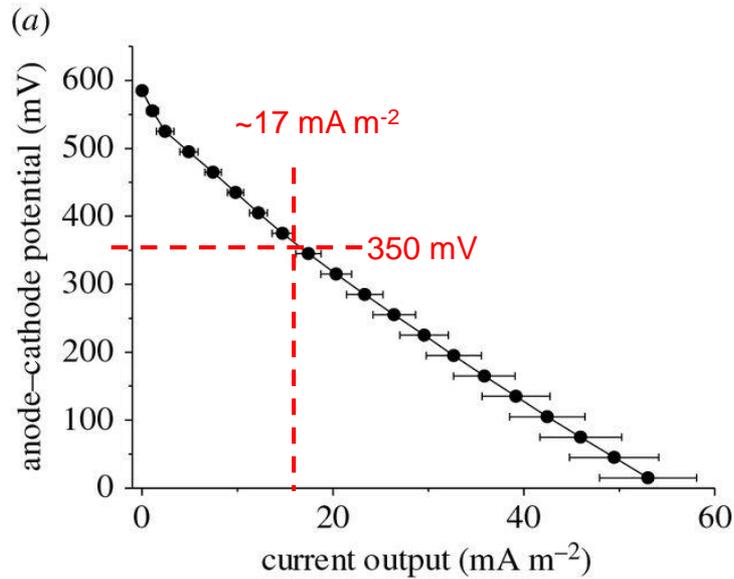


The environmental sensor requires:
5000 mC (@ 5V) per day

Photosynthetic-BES used to run environmental sensor in remote locations (e.g., tropical forest)



The environmental sensor requires:
5000 mC (@ 5V) per day

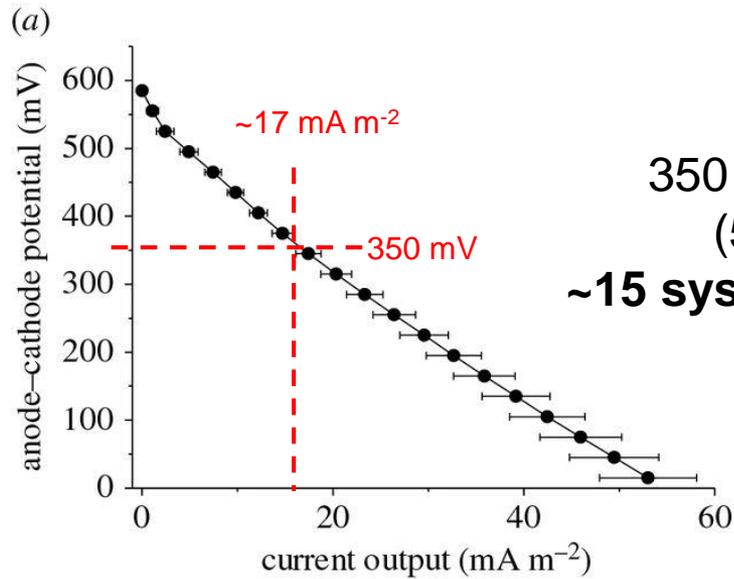


Royal Society journal *Open Science*, 2016 DOI: 10.1098/rsos.160249

Photosynthetic-BES used to run environmental sensor in remote locations (e.g., tropical forest)



The environmental sensor requires:
5000 mC (@ 5V) per day



Potential:
350 mV per system
(5V / 0.35V) =
~15 system to provide 5V

Royal Society journal *Open Science*, 2016 DOI: 10.1098/rsos.160249

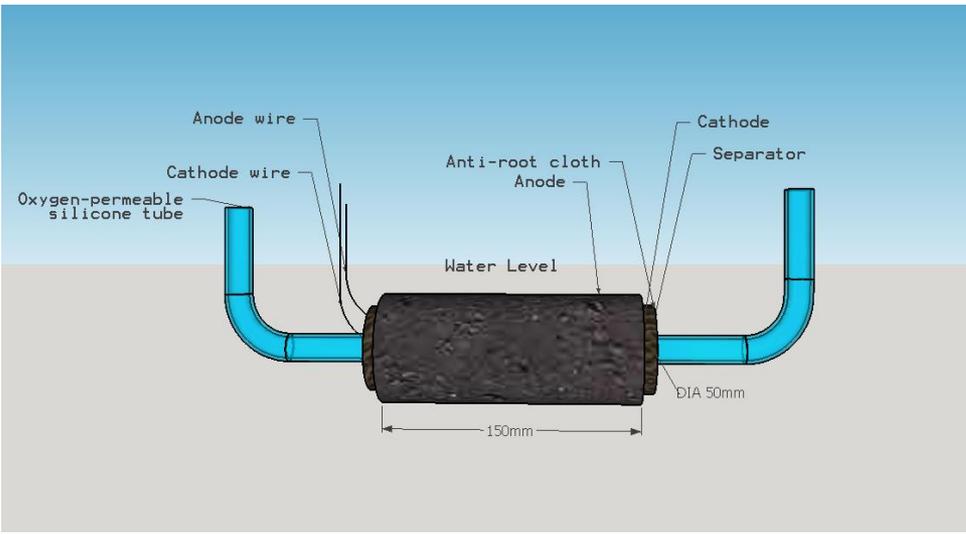
Current:
 $17 \text{ mA m}^{-2} = 1.7 \text{ } \mu\text{C s}^{-1} \text{ cm}^{-2}$
 $1.7 \text{ } \mu\text{C s}^{-1} \text{ cm}^{-2} = 146.9 \text{ mC cm}^{-2} \text{ per day}$
 $5000 / 146 = \sim \mathbf{34 \text{ cm}^{-2}}$

Photosynthetic-BES used to run environmental sensor in remote locations (e.g., tropical forest)



Artistic view

The real device might look like this



+ plants.....

In conclusion

photosynthetic-BES could be a feasible tool to:

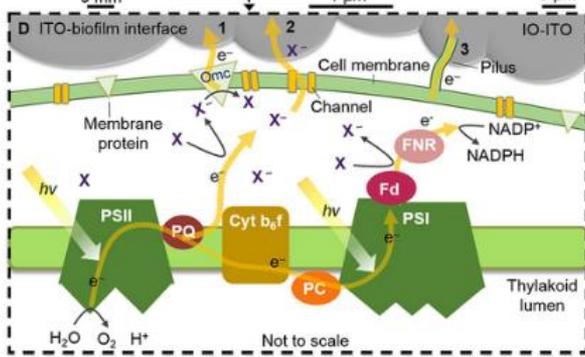
power
environmental
sensor in
remote
locations
(e.g.,
tropical
forest)



In conclusion

photosynthetic-BES could be a feasible tool to:

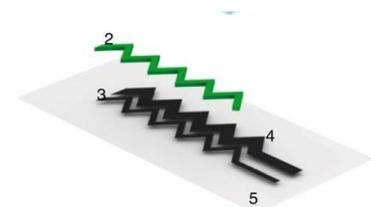
Tool of lab investigation



power
environmental
sensor in
remote
locations
(e.g.,
tropical
forest)



power
autonomous
 μ -chip
arm



Develop
educational
tool-kit



Be a living
sensor for
agricultural
application

Combined with
wastewater
treatment



Acknowledgments

Howe's lab



- **Chris Howe (Group Leader)**
- Jack Hervey (Graduate Student)
- Isabel Nimmo (Postdoctoral Researcher)
- Ellen Nisbet (Senior Research Associate)
- Elfadil Osman (Graduate Student)
- Stephen Rowden (Graduate Student)
- Barnaby Slater (Graduate Student)
- Laura Wey (Graduate Student)
- Wendy Gibson (Lab Manager)

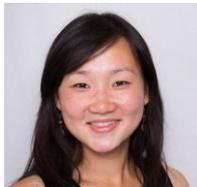
Acknowledgments



Chris Howe



Jenny Zhang



Bill Rutherford



Andrea Fantuzzi



Peter Nixon



Marin Sawa



Alison Smith



Erwin Reisner



Alasdair Davies



Rachael Kemp



Tuomas Knowles



Priyanka Jamwal



Lara Allen



Sue Harrison



Anand Savanth

Emre Ozer

Acknowledgments

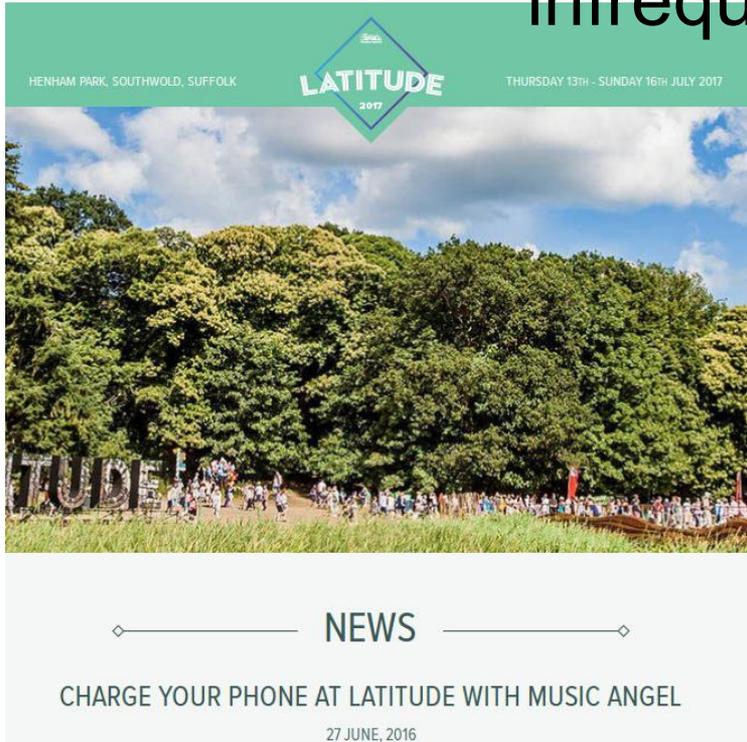


Acknowledgments

Thank you for your attention



Photosynthetic-BES to power small device in “infrequent” location/event



Music Angel Social Power: Li - ion battery (@5v)

Music Angel Mobile Charging are at the festival this year to keep you topped up with power. For £20 on pre-order, you get a Social Power battery that comes with an unlimited charging wristband – meaning you can swap the battery as many times as you like over the festival.

It takes 2 hours to charge an iPhone 6. For this phone, the battery has a voltage of 3.82V and the capacity is 1810 mAh. For a total of 6.91Wh.

At the festival the charging station are open from 9am to 11pm (14h). Therefore we could estimate up to $(14/2)= 7$ cycles to charge per day.

Therefore, the total electrical energy consumed per day could be up to $(6.91 \times 7)= 0.048$ kWh per day.

Latitude operates for 4 days, therefore at most you could consume $(0.048 \times 4)= 0.194$ kWh

During the festival, you will pay the electrical energy to charge your mobile $(20/0.194)=$ £ 103 per kWh

<http://www.musicangelsocialpower.com/>

<http://www.latitudefestival.com/news/charge-your-phone-latitude-music-angel>



To prove that photosynthetic organisms generate oxygen (and therefore electrons)



Oxygen electrode

Synechocystis
wt 6803

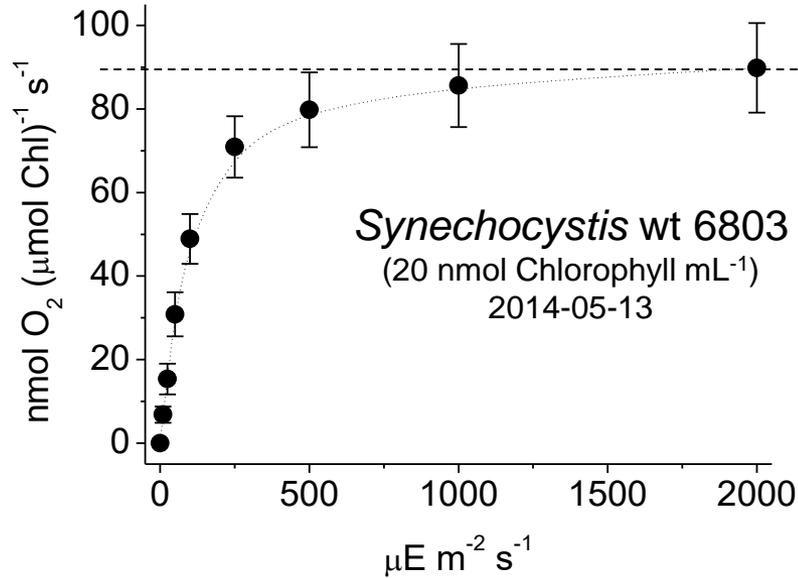
Ø: 1 cm

Top area : 0.79 cm²

h: 2.5 cm (for 2 mL)

Front area: 2.5 cm²

~90 nmol O₂ (μmol Chl)⁻¹ s⁻¹ (@ 2000 μE m⁻² s⁻¹)



20 nmol Chlorophyll mL⁻¹ x 15 mL = 300 nmol Chlorophyll

Photosynthetic oxygenic activity: ~90 nmol O₂ (μmol Chl)⁻¹ s⁻¹

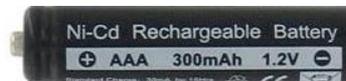
(90 nmol O₂ (μmol Chl)⁻¹ s⁻¹) x 300 nmol Chlorophyll = 27 μmol O₂ s⁻¹

Given: H₂O → O₂ + 4H⁺ + 4e⁻ (27 μmol O₂ s⁻¹) x 4 = 108 μmol e⁻ s⁻¹

Given: Coulomb = charge x F 108 μmol e⁻ s⁻¹ x F = 10.4 mC s⁻¹

= 10.4 mA

~ 250 mAh day⁻¹



Ni-Cd AAA (300 mAh)

pBES in school



**Hills Road
Sixth Form College**
Cambridge

Hills road sixth form college (Cambridge)
<http://www.hillsroad.ac.uk/>



**University
Technical
Colleges®**

<http://www.utoncollege.com/>
Karl Frearson Head of Science
k.frearson@etoncollege.org.uk

<http://www.utoncollege.com/>
[University Technical
College \(Cambridge\)](http://www.utoncollege.com/)
Alistair Easterfield
aeasterfield@utccambridge.co.uk



<http://www.etoncollege.com/>
Karl Frearson Head of Science
k.frearson@etoncollege.org.uk



**THE ENGLISH INTERNATIONAL SCHOOL
PRAGUE**
A NORD ANGLIA EDUCATION SCHOOL

Marketa Cechova (Prague, Czech Republic).
<http://www.nordangliaeducation.com/our-schools/prague>
Marketa.Cechova@eisp.cz



**Marlboro
College**

Marlboro College (USA)
(self-directed liberal arts education)
<https://www.marlboro.edu/>
danielm@marlboro.edu



KAOHSIUNG AMERICAN SCHOOL

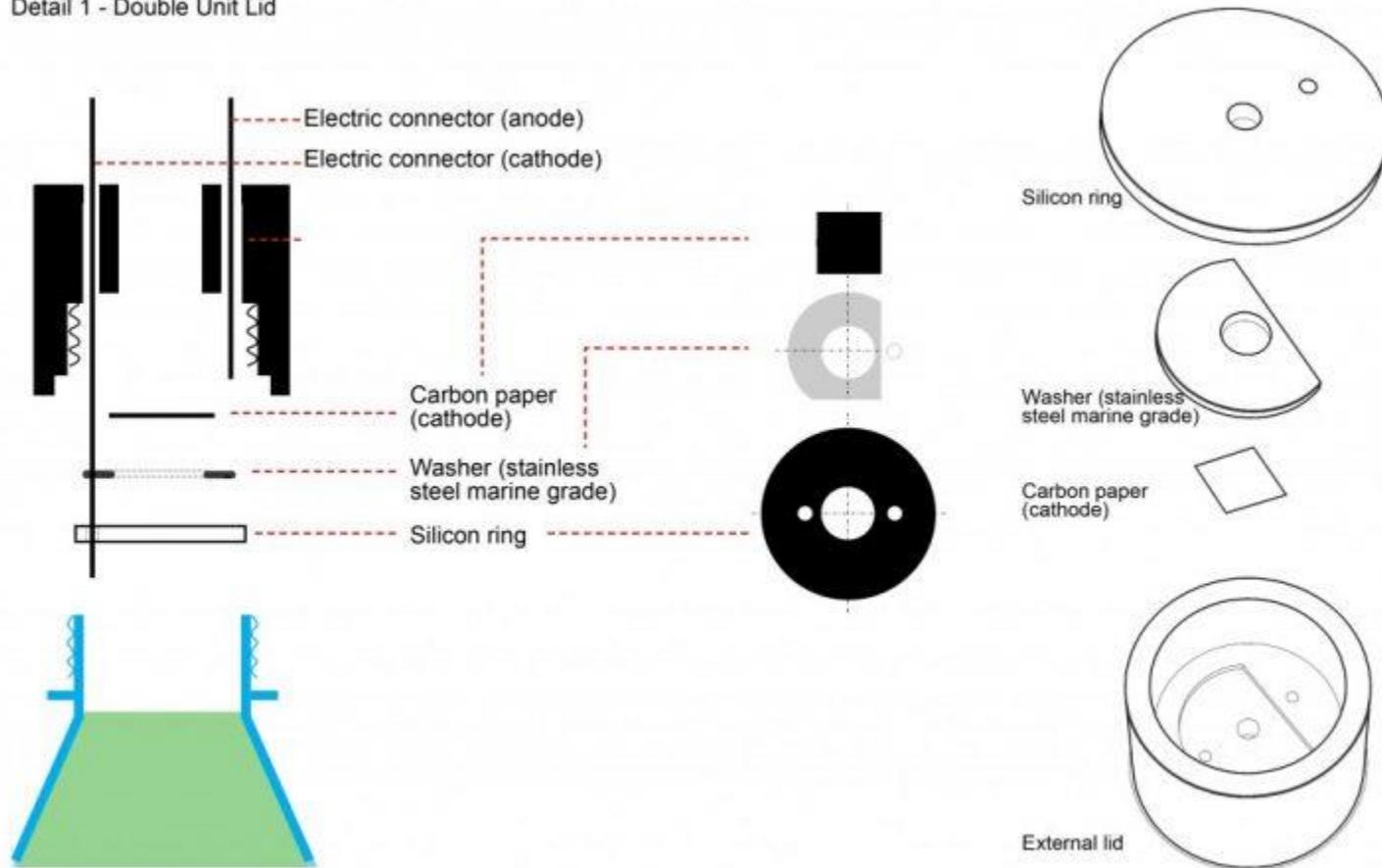
Jeff Roth-Vinson
Kaohsiung School, Taiwan
jrothvinson@kas.kh.edu.tw
<http://www.kas.tw/>

Filipa de Vilhena
Mais informados, mais capazes, mais livres.

Filipa de Vilhena HighSchool in Oporto, Portugal.
<http://www.filipa-vilhena.edu.pt/>
João Campos Grancho jcgrancho@gmail.com

pBES as educational tool-kit

Detail 1 - Double Unit Lid



<http://www.iaacblog.com/programs/bio-bottle-voltaic-bbv/>

pBES as educational tool-kit



1 *Article*

2 **Electrochemical characterisation of BBV systems**
3 **operated with algae and built with recycled materials**

4 Peter Bateson¹, Jack E.H. Fleet², Anthony S. Riseley², Elena Janeva³, Anastasia S. Marcella³,
5 Chiara Farinea³, Maria Kuptsova³, Núria Conde⁴, Christopher J. Howe², Paolo Bombelli^{2*} and
6 Brenda M. Parker¹

7 ¹ Dept. of Biochemical Engineering, UCL Bernard Katz Building, London WC1E 0AH

8 ² Department of Biochemistry, University of Cambridge, Hopkins Building, Downing Site, CB2 1QW

9 ³ Institute for Advanced Architecture of Catalonia, Pujades 102, Poble Nou, Barcelona 08005, Spain

10 ⁴ ICREA-Complex Systems Laboratory, Universitat Pompeu Fabra (UPF), Barcelona, Spain, Cell Signaling
11 Research Group, Departament de Ciències Experimentals i de la Salut, Universitat Pompeu Fabra (UPF),
12 Barcelona, Spain

13 * Correspondence: pb346@cam.ac.uk; Tel.: +44-01223-333684

Plant-BES in school

Filipa de Vilhena
Mais informados, mais capazes, mais livres.

Filipa de Vilhena HighSchool in Oporto, Portugal.

<http://www.filipa-vilhena.edu.pt/>

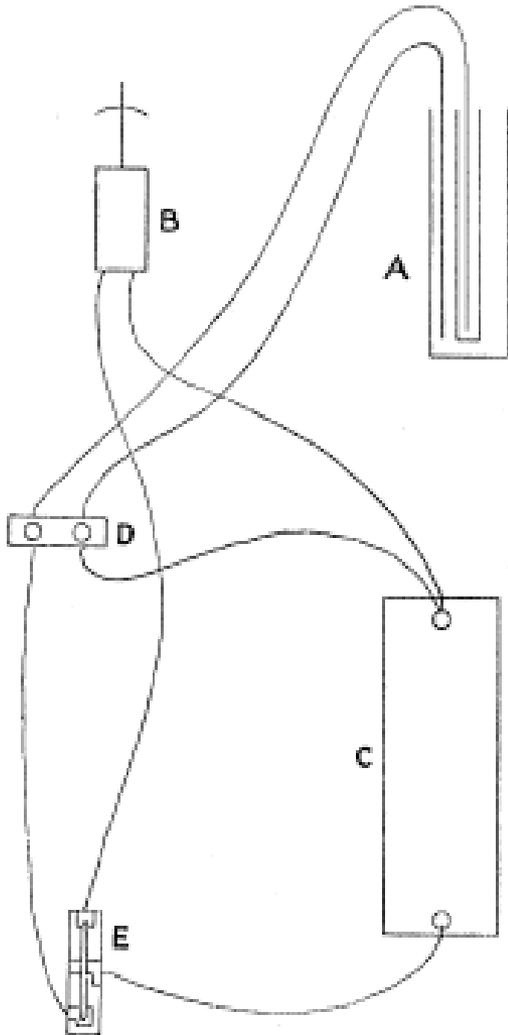
João Campos Grancho jcgrancho@gmail.com



CÉLULA	DIFERENÇA DE POTENCIAL / mV			
	16 maio	17 maio	18 maio	19 maio
controlo	109,4	108	100,4	69,6
1	447	503	504,2	501,2
2	239	386,2	302,8	303
3	269	249,1	250,4	249,3
4	183,4	225,6	226,3	227
5	178,5	140	196,4	195,3
6	171,9	161	172,1	172,2
7	189,1	194,2	200,2	201,3
8	177,7	172,1	178,1	184
9	282,4	302,3	304	304
10	190,2	201,2	203	213,4

Bio Electrochemical System (BES): is this concept new?

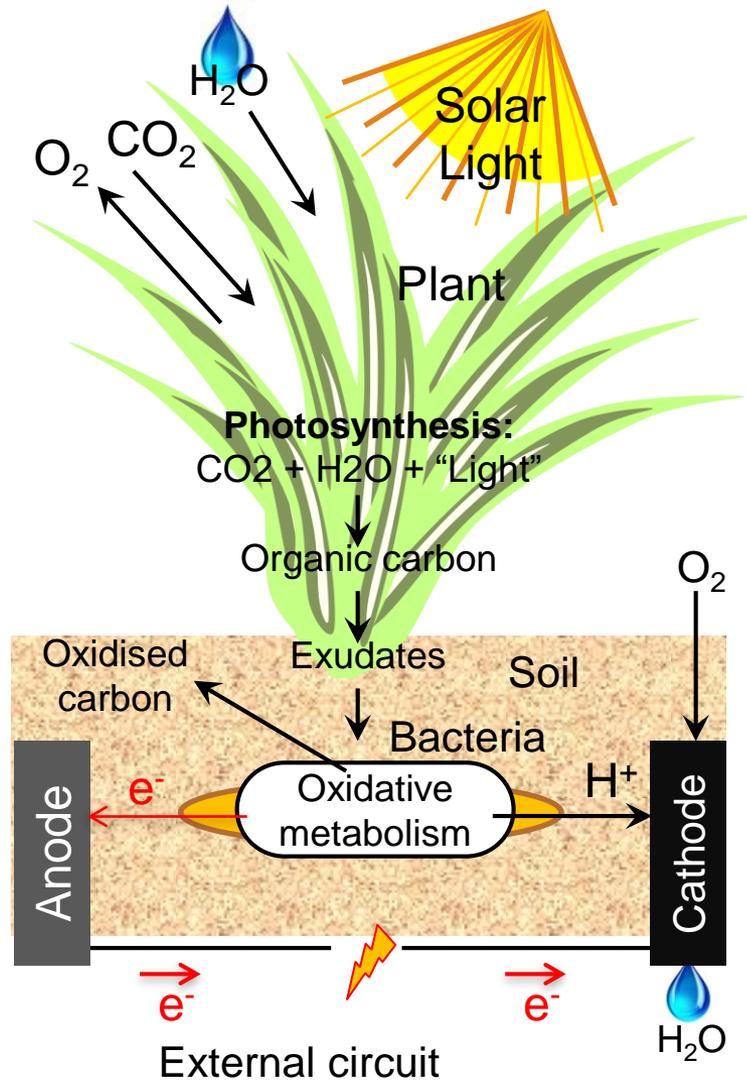
Proceedings of the Royal Society
of London. Vol. 84, N.571
(**Sep. 14, 1911**) pp.260-276

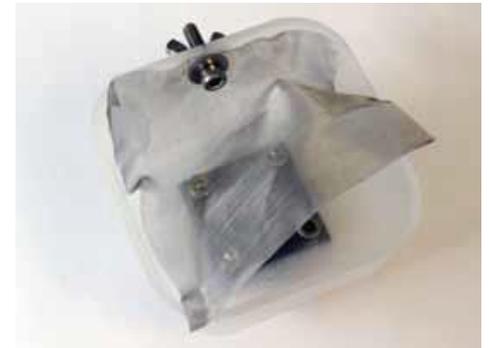
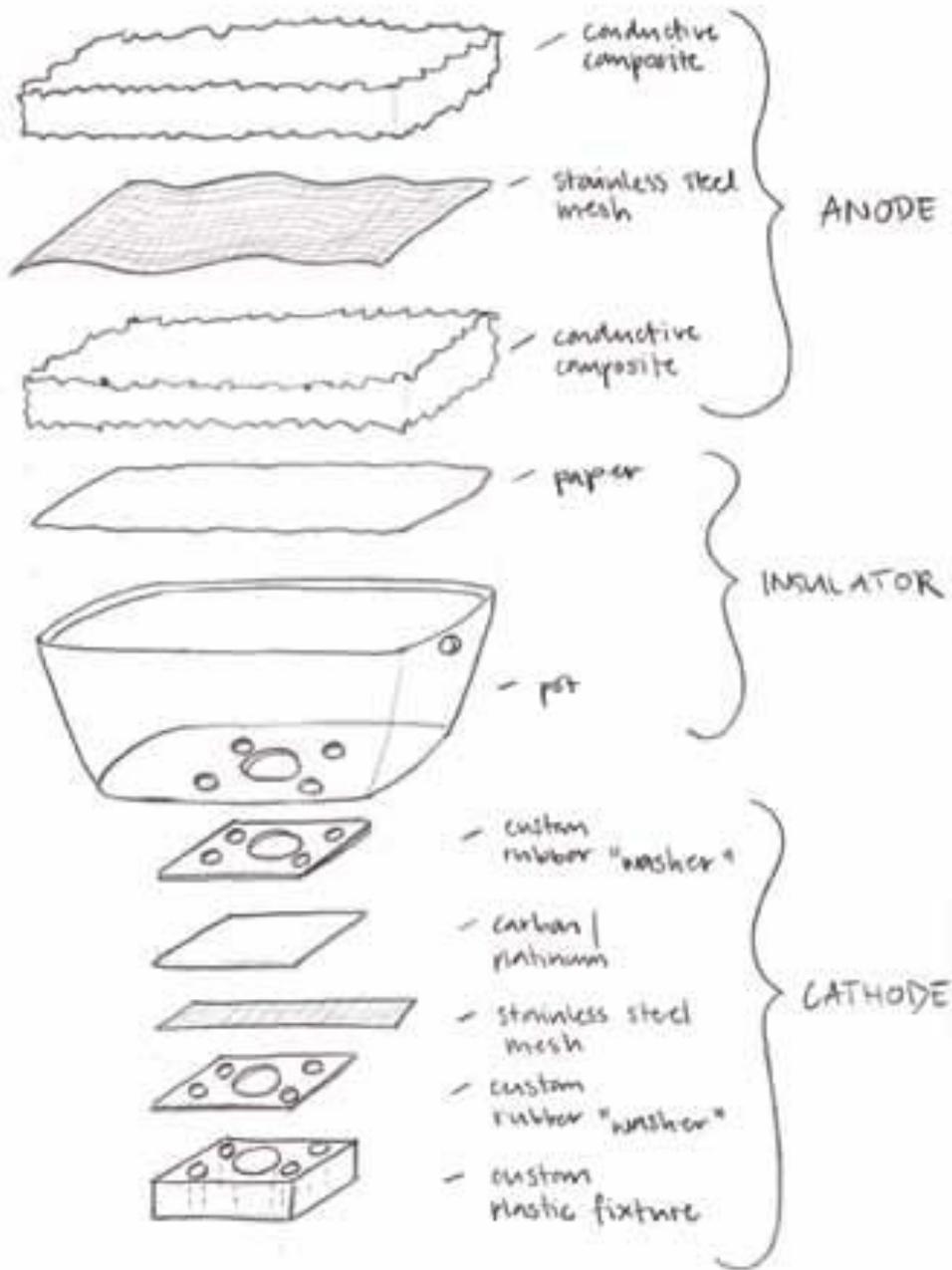


A, The cell.
B, Galvanometer.
C, Condenser.
D, Mercury cups
facilitate connec
with different ce
E, Morse key.

M. C. Potter
University of Durham
(Professor of botany)

Electrons from plants: how does it work?





Electrons from plants: what could we power?



~ 2000 W

Electrons from plants: what could we power?

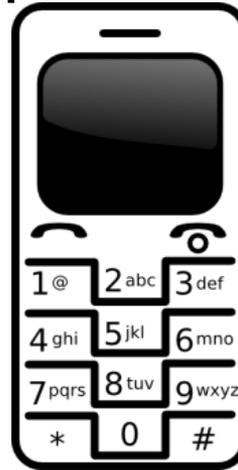
Environmental sensors
(e.g., humidity, temperature, light)



1-50 mW

Indeed

Mobile phones



100-500 mW

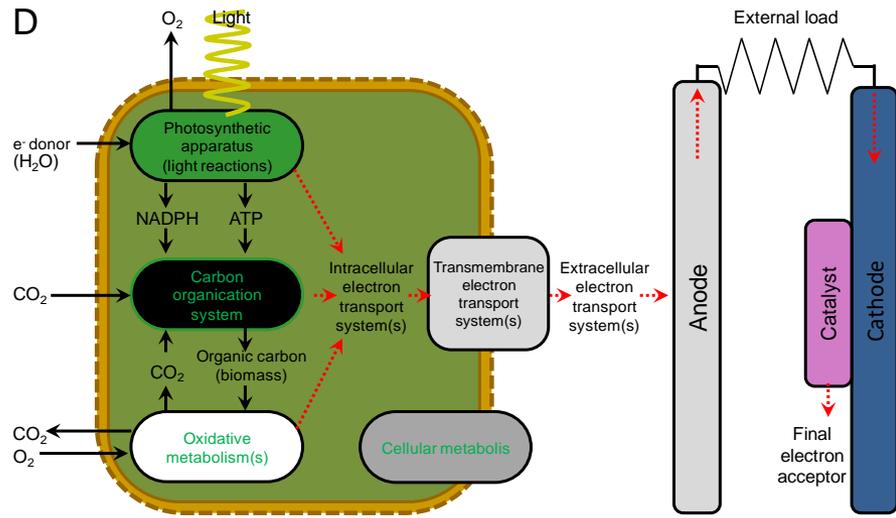
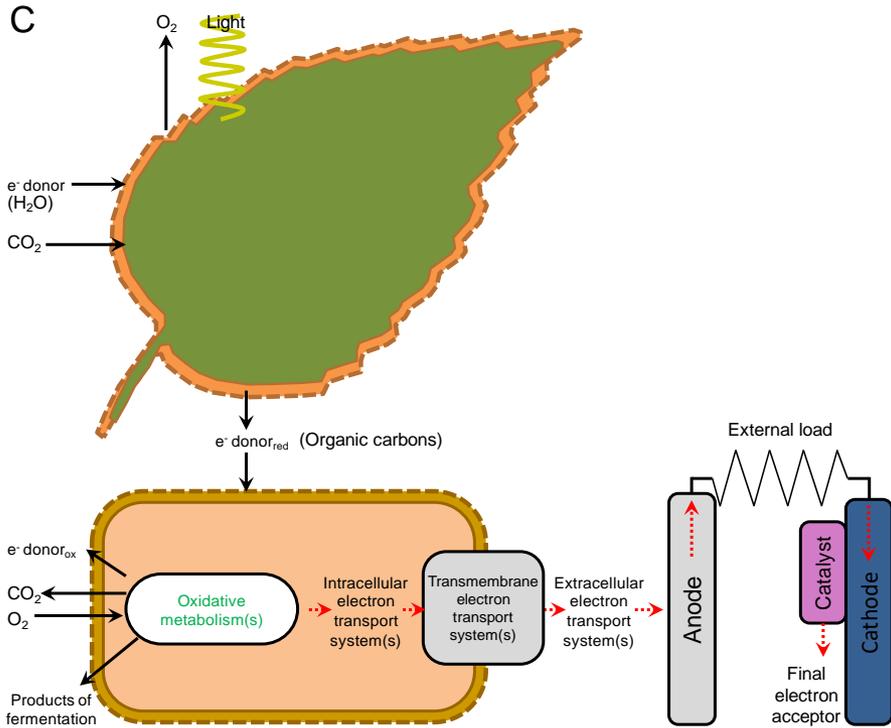
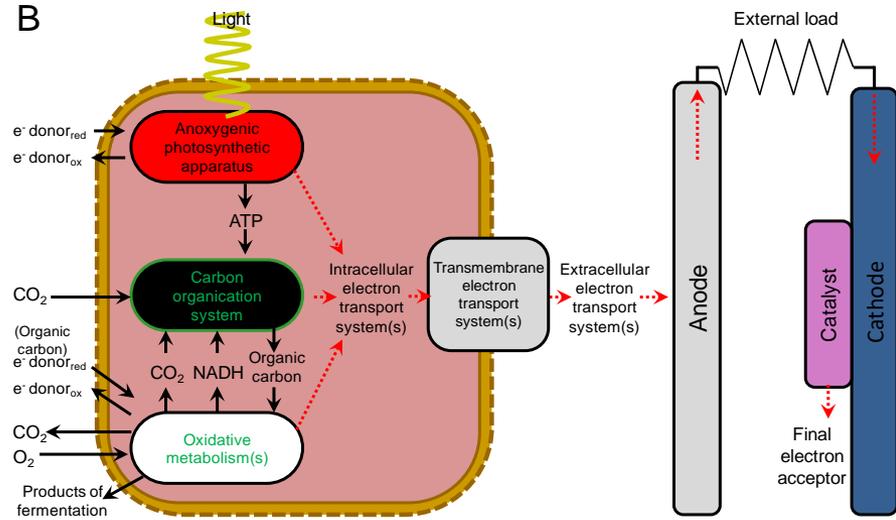
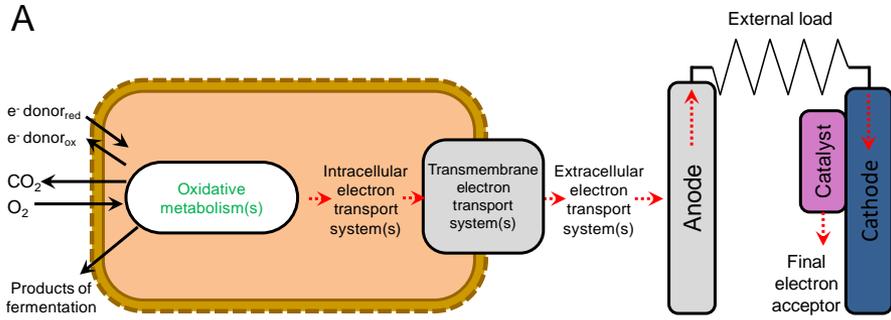
Almost there...

PCs



1-10 W

Maybe...



BES in the context of PV and FC

